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## **RESEARCH PAPER**

# Effect of preharvest application of different chemicals and plant growth regulators on biochemical parameters of mango (*Mangifera indica* L.) var. Amrapali

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**Abstract :** The present investigation on effect of preharvest application of different chemicals and plant growth regulators on biochemical parameters of mango (*Mangifera indica* L.) var. Amrapali was carried out during summer season of the year 2016 at Anand. The plants of mango were preharvest sprayed with  $CaCl_2 1 \%$ ,  $CaCl_2 2 \%$ ,  $Ca(NO_3)_2 1 \%$ ,  $Ca(NO_3)_2 2 \%$ ,  $KNO_3 1 \%$ ,  $KNO_3 2 \%$ ,  $GA_3 25 \text{ mg/l}$ ,  $GA_3 50 \text{ mg/l}$ , ethrel 0.1 ml/l and ethrel 0.2 ml/l. There were eleven treatment embedded in Completely Randomized Design replicated thrice. Thirty three uniform size tree of mango were selected. Fresh and mature fruits were harvested from trees which was preharvest sprayed with different chemicals and plant growth regulators and stored under ambient storage condition. Among all the treatment  $CaCl_2 2 \%$  recorded significantly highest total soluble solids, reducing sugar, total sugar, non-reducing sugar, ascorbic acid and minimum acidity consistently at harvest and every three days interval upto last ripening stage *i.e.* 4<sup>th</sup> day, 8<sup>th</sup> day, 12<sup>th</sup> day and 16<sup>th</sup> day under ambient storage condition.

Key Words : CaCl<sub>2</sub>, GA<sub>3</sub>, Fruit quality

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### **INTRODUCTION**

Mango (*Mangifera indica* L.) is a premier fruit crop of India. 'Indian mango' is the only mango tree commonly cultivated in many tropical and subtropical regions (Bose *et al.*, 2001). Fruit quality at harvest depends on the combined net effect of energy, water and flow into and out of the fruit. The same principles apply to post harvest quality changes, since continued respiration and transpiration constitute net losses of energy and water from the harvested fruit that result in quality loss. This quality should be increased by various chemicals and plant growth regulators applied at certain period before harvesting of fruits for proper maintaining the balance of nutrient in the fruit so that fruit quality and shelf-life should be maintained for longer duration after harvest (Shewfelt and Prussia, 2009). Among various chemicals and plant growth regulators, calcium is known to be essential plant nutrient involved in a number

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of physiological processes concerning membrane structure, function and enzyme activity (Jones and Lunt, 1967). Calcium has received considerable attention in recent year due to its desirable effect in delaying ripening and senescence, increasing firmness, vitamin C and phenolic content, reduce respiration, extending storage life and reducing the incidence of physiological disorder and storage rots. Calcium chloride treatment to fruits protects against post-harvest deterioration by binding with hydrolysis such as galacturonase and promotes shelf life. Calcium is essential for structural integrity of both the cell wall and plasma membranes. Calcium treatments have known to delay softening and improve the fruit quality.

#### MATERIAL AND METHODS

The experiment was conducted on effect of preharvest application of different chemicals and plant growth regulators on biochemical parameters of mango (*Mangifera indica* L.) var. Amrapali was carried out at Horticultural Research Farm and P. G. Laboratory, Department of Horticulture, B. A. College of Agriculture, Anand Agricultural University, Anand during summer season of the year 2016. There were eleven treatments embedded in Completely Randomized Design replicated thrice. Thirty three uniform size tree of mango var. Amrapali were selected and preharvest sprayed with different chemicals (CaCl<sub>2</sub> 1 %, CaCl<sub>2</sub> 2 %, Ca(NO<sub>3</sub>)<sub>2</sub> 1 %, Ca(NO<sub>3</sub>)<sub>2</sub> 2 %, KNO<sub>3</sub> 1 % and KNO<sub>3</sub> 2 %) and plant growth regulators (ethrel 0.1 ml/l and ethrel 0.2 ml/l) and control at twenty days before anticipated date of harvest while, GA<sub>3</sub> 25 mg /l and GA<sub>3</sub> 50 mg /l were sprayed at marble stage. The mature and uniform sized fruits were harvested from the respective trees and kept in ambient storage condition where observations were recorded regarding the biochemical parameters of the fruits. The details of the treatments applied in the present investigation are T<sub>1</sub>: CaCl<sub>2</sub> 1 %, T<sub>2</sub>: CaCl<sub>2</sub> 2 %, T<sub>3</sub>: Ca(NO<sub>3</sub>)<sub>2</sub> 1 %, T<sub>4</sub>: Ca(NO<sub>3</sub>)<sub>2</sub> 2 %, T<sub>5</sub>: KNO<sub>3</sub> 1 %, T<sub>6</sub>: KNO<sub>3</sub> 2 %, T<sub>7</sub>: ethrel 0.1 ml/l, T<sub>8</sub>: ethrel 0.2 ml/l, T<sub>9</sub>: GA<sub>3</sub> 25 mg /l, T<sub>10</sub>: GA<sub>3</sub> 50 mg /l and T11: Control (No spray).

#### **RESULTS AND DISCUSSION**

The results obtained from the present investigation was conducted on the preharvest application of different chemicals and PGRs influenced on mango fruit under ambient storage condition are presented in Table 1 to 3.

The chemicals and plant growth regulators significantly influenced on total soluble solids (°Brix) of mango at harvest and upto last ripening stage. Among all treatments,  $CaCl_2 2\%$  recorded significantly maximum total soluble solids consistently at harvest (7.72 °Brix), 4<sup>th</sup> day (16.00 °Brix), 8<sup>th</sup> day (20.97 °Brix), 12<sup>th</sup> day (21.68

(%) of mang	o var. Amrapa	ali									
Treatments		Total	soluble solid	Reducing sugar (%)							
		Sto	rage period (	Days)		Storage period (Days)					
	At harvest	4 <sup>th</sup>	8 <sup>th</sup>	$12^{th}$	16 <sup>th</sup>	At harvest	$4^{th}$	8 <sup>th</sup>	12 <sup>th</sup>	16 <sup>th</sup>	
T1: CaCl2 1 %	6.80 <sup>cd</sup>	14.53 <sup>cd</sup>	18.64 <sup>bc</sup>	20.77 <sup>bc</sup>	17.70 <sup>ab</sup>	3.14 <sup>b</sup>	5.85 <sup>de</sup>	7.49 <sup>a</sup>	3.09 <sup>b</sup>	2.96 <sup>b</sup>	
T <sub>2</sub> : CaCl <sub>2</sub> 2 %	7.72 <sup>a</sup>	16.00 <sup>a</sup>	20.97 <sup>a</sup>	21.68 <sup>a</sup>	17.75 <sup>a</sup>	3.49 <sup>a</sup>	6.65 <sup>a</sup>	7.51 <sup>a</sup>	3.62 <sup>a</sup>	3.12 <sup>a</sup>	
T <sub>3</sub> : Ca (NO <sub>3</sub> ) <sub>2</sub> 1%	7.55 <sup>a</sup>	13.55 <sup>e</sup>	18.22 <sup>cd</sup>	19.83 <sup>d</sup>	17.37 <sup>ab</sup>	3.08 <sup>b</sup>	$5.62^{\mathrm{f}}$	6.92 <sup>d</sup>	3.04 <sup>bc</sup>	2.95 <sup>b</sup>	
T <sub>4</sub> : Ca (NO <sub>3</sub> ) <sub>2</sub> 2%	6.05 <sup>e</sup>	13.55 <sup>e</sup>	17.87 <sup>d</sup>	19.00 <sup>e</sup>	16.00 <sup>d</sup>	3.44 <sup>a</sup>	6.00 <sup>cd</sup>	$6.29^{\mathrm{f}}$	2.71 <sup>c</sup>	2.66 <sup>c</sup>	
T <sub>5</sub> : KNO <sub>3</sub> 1 %	6.77 <sup>cd</sup>	15.33 <sup>ab</sup>	$20.88^{a}$	21.00 <sup>b</sup>	16.12 <sup>d</sup>	3.09 <sup>b</sup>	6.11 <sup>bc</sup>	6.16 <sup>g</sup>	2.51 <sup>d</sup>	2.42 <sup>d</sup>	
T <sub>6</sub> : KNO <sub>3</sub> 2 %	6.60 <sup>d</sup>	12.08 <sup>g</sup>	18.53 <sup>bc</sup>	20.35 <sup>cd</sup>	17.10 <sup>abc</sup>	3.44 <sup>a</sup>	5.80 <sup>e</sup>	7.35 <sup>bc</sup>	3.03 <sup>b</sup>	2.91 <sup>b</sup>	
T7: Ethrel 0.1 ml/l	7.20 <sup>b</sup>	14.94 <sup>bc</sup>	18.98 <sup>b</sup>	20.67 <sup>bc</sup>	17.10 <sup>abc</sup>	3.47 <sup>a</sup>	6.26 <sup>b</sup>	7.22 <sup>c</sup>	3.50 <sup>a</sup>	3.11 <sup>a</sup>	
T <sub>8</sub> : Ethrel 0.2 ml/l	6.75 <sup>cd</sup>	15.97 <sup>a</sup>	20.53 <sup>a</sup>	21.05 <sup>b</sup>	16.50 <sup>cd</sup>	3.49 <sup>a</sup>	6.62 <sup>a</sup>	7.39 <sup>ab</sup>	3.58 <sup>a</sup>	2.93 <sup>b</sup>	
T <sub>9</sub> : GA <sub>3</sub> 25 mg/l	6.60 <sup>d</sup>	14.18 <sup>de</sup>	17.95 <sup>d</sup>	$18.45^{\mathrm{f}}$	17.05 <sup>bc</sup>	2.33 <sup>d</sup>	5.87 <sup>de</sup>	6.71 <sup>e</sup>	2.56 <sup>d</sup>	2.55 <sup>c</sup>	
T10: GA3 50 mg/l	6.92 <sup>bc</sup>	12.83 <sup>f</sup>	16.53 <sup>e</sup>	18.87 <sup>ef</sup>	17.51 <sup>ab</sup>	2.81 <sup>c</sup>	5.81 <sup>e</sup>	$6.24^{\mathrm{fg}}$	3.16 <sup>b</sup>	3.00 <sup>ab</sup>	
T <sub>11</sub> : Control	6.20 <sup>e</sup>	10.87 <sup>h</sup>	$15.71^{\mathrm{f}}$	17.95 <sup>g</sup>	15.33°	2.34 <sup>d</sup>	4.76 <sup>g</sup>	6.15 <sup>g</sup>	3.61 <sup>a</sup>	2.64 <sup>c</sup>	
S.E. ±	0.09	0.21	0.16	0.16	0.20	0.04	0.05	0.04	0.04	0.04	
F-test	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	
C V %	216	2 65	1 49	1 37	2.01	196	1 55	1.08	2.18	2.60	

 Table 1: Effect of preharvest application of different chemicals and plant growth regulators on total soluble solids ("Brix) and reducing sugar (%) of mango var. Amrapali

Note: Treatment means with the letter/letters in common are not significantly different by Duncan's new multiple range test at 5 % level of significance

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°Brix) and 16th day (17.75°Brix) under ambient storage conditions. The total soluble solid increased slowly during storage period upto 12th day and declined thereafter. It may be due to the increased in soluble sugars content and total soluble sugars caused by hydrolysis of polysaccharides like starch, cellulose and pectin substances in to simpler substances. The change in total soluble solids during storage period might be a result of the transformation of organic compound into total soluble solids throughout the enzymatic activities and reaction under calcium effect (Karemera and Habimana, 2014). A similar view was also shared by Kirmani et al. (2013) in plum and Yadav et al. (2009) in ber fruit.

There were significant effect of various chemicals and plant growth regulators observed on reducing sugar (%) of mango at harvest and every three days interval up to last ripening stage. CaCl, 2 % and ethrel 0.2 ml/l treatment recorded significantly maximum reducing sugar (%) at harvest (3.49 %) while, CaCl<sub>2</sub> 2% treatment obtained significantly maximum reducing sugar (%) at 4<sup>th</sup> day (6.65 %), 8<sup>th</sup> day (7.51%), 12<sup>th</sup> day (3.62%) and at 16<sup>th</sup> day (3.12%) as compared to rest of the treatments. The accumulation of reducing sugar was gradually increased in fruits preharvest treated with chemicals and growth regulators with a slight decline at the end of storage periods and it was significantly highest with CaCl<sub>2</sub> 2 (%) at harvest, 4<sup>th</sup> day, 8<sup>th</sup> day, 12<sup>th</sup> day and at 16<sup>th</sup> day as compared to rest of the treatments. The initial increase in reducing sugar might be due to the conversion of starch into reducing sugar and later on reduction could be possible due to utilization of sugar in the process of respiration. The percentage of reducing sugar increased slowly during storage period upto 8th day and declined thereafter. The increase in reducing sugar might be due to increased rate of starch degradation by amylase activity. The present findings are in agreement with Singh et al. (2007) in strawberry.

Total sugar (%) of mango were significantly influenced by various chemicals and plant growth regulators mango at harvest and upto last ripening stage. Among all treatments, CaCl<sub>2</sub> 2 per cent recorded significantly maximum total sugar consistently at harvest (6.30 %), 4<sup>th</sup> day (15.43 %), 8<sup>th</sup> day (18.57 %), 12<sup>th</sup> day (18.95%) and  $16^{\text{th}}$  day (18.27%) during storage period. The percentage of total sugar increased slowly during storage period upto 12th day and declined thereafter and total sugar was found to increase in calcium chloride treated fruits might be due to slowly conversion to sugar on storage reported by Bhalerao et al. (2010) in sapota. Whereas, calcium pectate is important components of the cell wall. Therefore, adequate amount of calcium may help to reduce conversion of acid into sugar reported by Lal et al. (2011) in apricot. Similar results were also reported Karemera and Habimana (2014) in mango,

mango var.	Amrapali										
Treatments		Non-reducing sugar (%)									
Treatments	Storage period (Days)					Storage period (Days)					
	At harvest	4 <sup>th</sup>	8 <sup>th</sup>	$12^{th}$	16 <sup>th</sup>	At harvest	$4^{th}$	8 <sup>th</sup>	12 <sup>th</sup>	16 <sup>th</sup>	
T1: CaCl2 1 %	5.78 <sup>b</sup>	14.34 <sup>b</sup>	17.82 <sup>c</sup>	17.93 <sup>c</sup>	17.97 <sup>b</sup>	2.64 <sup>ab</sup>	8.49 <sup>a</sup>	10.33 <sup>cd</sup>	14.84 <sup>b</sup>	15.01 <sup>a</sup>	
T <sub>2</sub> : CaCl <sub>2</sub> 2 %	6.30 <sup>a</sup>	15.43 <sup>a</sup>	18.57 <sup>a</sup>	18.95 <sup>a</sup>	18.27 <sup>a</sup>	2.81 <sup>a</sup>	$8.78^{a}$	11.06 <sup>a</sup>	15.33ª	15.15 <sup>a</sup>	
T <sub>3</sub> : Ca(NO <sub>3</sub> ) <sub>2</sub> 1%	5.30 <sup>c</sup>	12.66 <sup>d</sup>	17.48 <sup>d</sup>	17.59 <sup>cd</sup>	16.96 <sup>cd</sup>	2.22 <sup>c</sup>	7.04 <sup>d</sup>	10.56 <sup>bc</sup>	14.55 <sup>b</sup>	14.01 <sup>c</sup>	
T <sub>4</sub> : Ca(NO <sub>3</sub> ) <sub>2</sub> 2%	6.24 <sup>a</sup>	13.71 <sup>c</sup>	17.28 <sup>de</sup>	17.44 <sup>d</sup>	17.20 <sup>c</sup>	2.80 <sup>a</sup>	7.71 <sup>c</sup>	10.99 <sup>a</sup>	14.73 <sup>b</sup>	14.54 <sup>b</sup>	
T <sub>5</sub> : KNO <sub>3</sub> 1 %	5.84 <sup>b</sup>	12.78 <sup>d</sup>	16.17 <sup>f</sup>	16.46 <sup>e</sup>	16.06 <sup>e</sup>	2.75 <sup>a</sup>	6.67 <sup>d</sup>	10.01 <sup>d</sup>	13.95 <sup>cd</sup>	13.63 <sup>de</sup>	
T <sub>6</sub> : KNO <sub>3</sub> 2 %	6.23 <sup>a</sup>	13.56 <sup>c</sup>	$16.22^{\text{f}}$	16.59 <sup>e</sup>	16.29 <sup>e</sup>	2.79 <sup>a</sup>	7.77 <sup>°</sup>	$8.87^{\mathrm{f}}$	13.57 <sup>f</sup>	13.39 <sup>e</sup>	
T <sub>7</sub> : Ethrel 0.1 ml/l	6.27 <sup>a</sup>	15.04 <sup>a</sup>	18.23 <sup>b</sup>	$18.82^{a}$	$18.18^{ab}$	$2.80^{a}$	$8.78^{\mathrm{a}}$	11.00 <sup>a</sup>	15.32 <sup>a</sup>	15.07 <sup>a</sup>	
T <sub>8</sub> : Ethrel 0.2 ml/l	6.29 <sup>a</sup>	15.04 <sup>a</sup>	18.38 <sup>ab</sup>	18.34 <sup>b</sup>	18.03 <sup>ab</sup>	2.80 <sup>a</sup>	8.41 <sup>ab</sup>	10.99 <sup>a</sup>	14.76 <sup>b</sup>	15.10 <sup>a</sup>	
T <sub>9</sub> : GA <sub>3</sub> 25 mg/l	5.11 <sup>d</sup>	12.44 <sup>d</sup>	$15.98^{\text{f}}$	16.50 <sup>e</sup>	16.30 <sup>e</sup>	2.78 <sup>a</sup>	6.57 <sup>d</sup>	9.27 <sup>e</sup>	13.94 <sup>cde</sup>	13.74 <sup>d</sup>	
T10: GA3 50 mg/l	5.38 <sup>c</sup>	13.70 <sup>c</sup>	16.99 <sup>e</sup>	17.22 <sup>d</sup>	16.83 <sup>d</sup>	2.58 <sup>b</sup>	7.89 <sup>bc</sup>	10.75 <sup>ab</sup>	14.05 <sup>c</sup>	13.82 <sup>cd</sup>	
T <sub>11</sub> : Control	5.07 <sup>d</sup>	11.35 <sup>e</sup>	15.01 <sup>g</sup>	$15.47^{\mathrm{f}}$	13.21 <sup>f</sup>	2.72 <sup>ab</sup>	6.60 <sup>d</sup>	$8.86^{\mathrm{f}}$	11.86 <sup>g</sup>	$10.57^{\mathrm{f}}$	
S. E. ±	0.04	0.13	0.09	0.12	0.08	0.05	0.16	0.12	0.12	0.08	
F-test	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	
C. V. %	1.21	1.66	0.95	1.23	0.81	2.98	3.66	2.05	1.52	1.03	

Table 2: Effect of preharvest application of different chemicals and plant growth regulators on total sugar (%) and non-reducing sugar (%) of

Note: Treatment means with the letter/letters in common are not significantly different by Duncan's new multiple range test at 5 % level of significance

94 Internat. J. agric. Sci. | Jan., 2018 | Vol. 14 | Issue 1 | 92-96 Hind Agricultural Research and Training Institute Mahajan and Sharma (2000) in peach.

There were significant effect of various chemicals and plant growth regulators observed on non-reducing sugar (%) of mango at harvest and upto last ripening stage. CaCl<sub>2</sub> 2 % recorded significantly maximum nonreducing sugar (%) consistently at harvest (2.81 %), 4<sup>th</sup> day (8.78 %), 8th day (11.06 %), 12th day (15.33 %) and 16<sup>th</sup> day (15.15 %) as compared to rest of the treatments. It was gradually increases during storage period upto 12<sup>th</sup> day and declined thereafter. The increase in the nonreducing sugar might be due to the hydrolysis of starch and conversion in the pectin substances from water insoluble to water soluble fractions. The increase in nonreducing sugar during storage was due to the conversion of starch into sugar as reported by Bhalerao et al. (2010) in sapota. These results are in accordance with the findings of Karemera and Habimana (2014) in mango and Jawandha et al. (2007) in ber.

The chemicals and plant growth regulators were significantly influenced on ascorbic acid (mg/100 g pulp) content of mango at harvest and every three days interval upto last ripening stage. Among all treatments,  $CaCl_2 2$  per cent recorded significantly maximum ascorbic acid (mg/100 g pulp) consistently at harvest (51.54 mg/100 g pulp), 4<sup>th</sup> day (45.83 mg/100 g pulp), 8<sup>th</sup> day (42.26 mg/100 g pulp), 12<sup>th</sup> day (39.63 mg/100 g pulp) and 16<sup>th</sup> day (35.42 mg/100 g pulp). CaCl<sub>2</sub> 2 per cent treatment was

statistically superior over all other treatments. It was gradually decreased in fruits preharvest treated with chemicals and growth regulators at the end of storage periods. It might be due to calcium delayed the rapid oxidation of ascorbic acid. The highest retention of ascorbic acid during storage by calcium sprays might be due to continue synthesis of its precursor like Glucose-6-phosphate during conversion as starch into various sugars and slow rate of oxidation reported by Yadav *et al.* (2009) in ber. These results are in accordance with the finding of Jawandha *et al.* (2007) in ber and Lal *et al.* (2011) in apricot.

There were significant effect of various chemicals and plant growth regulators on acidity of mango at harvest and every three days interval up to last ripening stage. CaCl<sub>2</sub> 2 per cent treatment recorded significantly minimum acidity at harvest (0.352 %), 4<sup>th</sup> day (0.258 %), 8<sup>th</sup> day (0.199 %), 12<sup>th</sup> day (0.101 %) and 16<sup>th</sup> day (0.096 %) as compared to rest of the treatments. The decrease in the total titratable acidity might be due to increase in the total sugar content of the fruits. At the time of maturity, fruits will be having higher amount of acidity, but as the fruits advance towards ripening, acid content will decrease. These results are in line with findings of Ingle *et al.* (1982), who observed a decrease in acidity during ripening of sapota fruits. The acidity of mango fruits generally decreases with advancement of

of mango v	ar. Amrapali									
Treatments		Acidity (%)								
		rage period (I	Storage period (Days)							
	At harvest	4 <sup>th</sup>	8 <sup>th</sup>	12 <sup>th</sup>	16 <sup>th</sup>	At harvest	$4^{th}$	8 <sup>th</sup>	12 <sup>th</sup>	16 <sup>th</sup>
T1: CaCl2 1 %	45.53 <sup>cde</sup>	38.78 <sup>e</sup>	36.09 <sup>de</sup>	33.15 <sup>g</sup>	28.34 <sup>ef</sup>	$0.356^{\mathrm{f}}$	0.292 <sup>d</sup>	0.247 <sup>c</sup>	0.118 <sup>e</sup>	0.101 <sup>d</sup>
T <sub>2</sub> : CaCl <sub>2</sub> 2 %	51.54 <sup>a</sup>	45.83 <sup>a</sup>	42.26 <sup>a</sup>	39.63 <sup>a</sup>	35.42 <sup>a</sup>	$0.352^{\mathrm{f}}$	0.258 <sup>e</sup>	0.199 <sup>e</sup>	0.101 <sup>g</sup>	0.096 <sup>e</sup>
T <sub>3</sub> : Ca(NO <sub>3</sub> ) <sub>2</sub> 1%	46.11 <sup>cd</sup>	41.92 <sup>c</sup>	38.82 <sup>b</sup>	35.65°	31.23 <sup>c</sup>	0.411 <sup>d</sup>	0.390 <sup>a</sup>	0.278 <sup>a</sup>	0.177 <sup>a</sup>	0.122 <sup>a</sup>
T <sub>4</sub> : Ca(NO <sub>3</sub> ) <sub>2</sub> 2%	46.37 <sup>bc</sup>	41.28 <sup>c</sup>	37.43 <sup>c</sup>	34.85 <sup>de</sup>	29.39 <sup>de</sup>	0.380 <sup>e</sup>	0.360 <sup>b</sup>	0.251 <sup>bc</sup>	0.143 <sup>d</sup>	0.101 <sup>d</sup>
T <sub>5</sub> : KNO <sub>3</sub> 1 %	44.54 <sup>e</sup>	38.85°	36.44 <sup>d</sup>	$33.92^{\mathrm{f}}$	27.56 <sup>f</sup>	0.377 <sup>e</sup>	0.300 <sup>cd</sup>	0.214 <sup>d</sup>	$0.111^{f}$	$0.098^{de}$
T <sub>6</sub> : KNO <sub>3</sub> 2 %	46.35 <sup>bc</sup>	41.99 <sup>c</sup>	39.03 <sup>b</sup>	36.90 <sup>b</sup>	32.84 <sup>b</sup>	0.421 <sup>cd</sup>	0.306 <sup>c</sup>	0.201 <sup>e</sup>	0.102 <sup>g</sup>	$0.098^{de}$
T <sub>7</sub> : Ethrel 0.1 ml/l	45.01 <sup>de</sup>	38.77°	37.33°	34.23 <sup>ef</sup>	29.75 <sup>d</sup>	0.451 <sup>b</sup>	0.353 <sup>b</sup>	0.255 <sup>b</sup>	0.149 <sup>c</sup>	$0.098^{de}$
T <sub>8</sub> : Ethrel 0.2 ml/l	$42.34^{\mathrm{f}}$	37.68 <sup>f</sup>	35.52 <sup>ef</sup>	31.90 <sup>h</sup>	25.60 <sup>g</sup>	$0.354^{\mathrm{f}}$	0.292 <sup>d</sup>	0.200 <sup>e</sup>	$0.108^{f}$	0.097 <sup>e</sup>
T <sub>9</sub> : GA <sub>3</sub> 25 mg/l	45.49 <sup>cde</sup>	39.92 <sup>d</sup>	37.88 <sup>c</sup>	34.90 <sup>d</sup>	28.61d <sup>ef</sup>	$0.353^{\mathrm{f}}$	0.260 <sup>e</sup>	0.202 <sup>e</sup>	$0.108^{f}$	0.108 <sup>c</sup>
T10: GA3 50 mg/l	47.41 <sup>b</sup>	42.98 <sup>b</sup>	39.00 <sup>b</sup>	36.97 <sup>b</sup>	31.98 <sup>bc</sup>	0.431 <sup>c</sup>	0.351 <sup>b</sup>	0.256 <sup>b</sup>	0.147 <sup>c</sup>	0.101 <sup>d</sup>
T <sub>11</sub> : Control	43.10 <sup>f</sup>	37.65 <sup>f</sup>	$35.22^{\mathrm{f}}$	32.01 <sup>h</sup>	25.46 <sup>g</sup>	0.474 <sup>a</sup>	0.386 <sup>a</sup>	0.272 <sup>a</sup>	0.169 <sup>b</sup>	0.112 <sup>b</sup>
S.E. ±	0.34	0.28	0.22	0.20	0.37	0.005	0.003	0.002	0.001	0.001
F-test	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.	Sig.
C V %	1 27	1 18	1.03	1.01	2 17	1 97	1 54	1 75	171	1 23

 Table 3 : Effect of preharvest application of different chemicals and plant growth regulators on ascorbic acid (mg/100 g pulp) and acidity (%) of mango var. Amrapali

Note: Treatment means with the letter/letters in common are not significantly different by Duncan's New Multiple Range Test at 5 % level of significance

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storage period. Decrease in acidity might be attributed to conversion of acids into sugars during respiration. These results are accordance with the findings of Kardum (2004) in fig, Jawandha *et al.* (2007) in ber and Kaundal *et al.* (2000) in plum fruits.

#### **Conclusion:**

The results obtained from present investigation concluded that,  $CaCl_2$  2 per cent treatment recorded maximum total soluble solids, total sugar, reducing sugar, non-reducing sugar and ascorbic acid content of mango fruit while, minimize the acidity of fruit under ambient storage condition.

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