Relative toxicity of various nickel species on seed germination and early seedling growth of *Vigna unguiculata* L.

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Cowpea (*Vigna unguiculata* L.) seedlings were raised in water culture and exposed to varying concentrations $(1,10,100,1000\mu M)$ of nickel as Ni (NO₃)₂ NiSO₄ and NiCl₂ with a view to compare the effect of the above nickel species on seed germination, ultimate percentage germination, dry matter production, chlorophyll contents and soluble protein (leaf) contents. The ultimate germination was 100% (except in 1000 μ M) in all the treatments of various nickel species. The germination speed was found maximum at 1 μ M nickel concentration. Lower concentration of nickel resulted in an enhancement while higher levels resulted decrease in fresh mass, dry matter production and chlorophyll contents of seedlings. Soluble leaf protein contents increased linearly with increasing application rates of nickel. In general, NiCl₂ was more inhibitory than NiSO₄ and Ni (NO₃)₂. Ni (NO₃)₂ was found to be least toxic. The extent of inhibition increased with enhanced levels of Ni₂⁺ ions. The toxicity series was found to be NiCl₂ > NiSO₄ > Ni (NO₃)₂.

Key words : Nickel, Vigna unguiculata L. Toxicity, Early seedling growth.

INTRODUCTION

Heavy metal contaminants affect the biosphere in many places worldwide (Cunningham *et.al.*, 1997; Raskin and Ensley, 2000; Meagher, 2000). In the majority of natural environments the concentration of heavy metal in soil is low and does not cause any significant phytotoxic effects (Gratao et al., 2005). Some heavy metals such as Cu, Zn, and Ni are essential micronutrient for plants, but are toxic to organism at high concentrations (Munzuroglu and Geckil, 2002). Although essential heavy metal ions, such as Ni⁺², are of major importance in different enzymatic reactions, excess cellular levels of such metals are toxic to all living cells. Nickel has one essential role in plants, which is to form the hexameric enzyme urease (E.C. 3.5.1.5.3) (Gerendas and Settelmacher, 1999). Nickel is not toxic at low concentration, but it becomes toxic at high concentrations (Poulik, 1997).

Seed is a developmental stage in plant life cycle that is highly protected against various external stresses. However, soon after imbition and subsequent vegetative developmental processes, they become stress sensitive in general. Therefore, according to Karssen, 1982; Pritchard *et al.*, 1993; Bungard *et al.*, 1997 seeds are thought to carefully monitor such external parameters as light, temperature and nutrient in order to maintain the protective state until external conditions become favourable for following developmental processes. Seed germination represents an important and initial phase in the life cycle of plants (Bishnoi *et al.*, 1993). Seed germination and early seedling growth responses of plants to adverse environmental conditions are critical for raising a successful agricultural crop stand density and establishment of resultant crop especially under stress conditions (Jagetiya and Aery, 1994a). A number of environmental factors together with the make up of seed affect germination phenomenon. The subject has attracted the attention of many workers right from the dawn of scientific research. Many treatises, review and proceedings have produced voluminous finding on germinability of seeds of several plants.

Several studies have been conducted in order to evaluate the effect of different heavy metal concentration on living plants (Thompson *et al.*, 1997). Most of these studies have been conducted using seedling or adult plants (Flores Tana *et al.*, 1999; Lee and Leustek, 1999; Chatterjee and Chatterjee 2000; Gratton *et al.*, 2000; Oneel *et al.*, 2000; Pichtel *et al.*, 2000). In a few study the seeds have been exposed to the contaminants (Clair *et al.*, 1991; Vajtechova and Leblova, 1991; Xiong, 1998).

Investigations on the effect of nickel on seed germination, growth and crop yield have been given conflicting results (Mishra and Kar, 1974). Das *et al.* (1978) responded that rice seeds when treated with nickel salts increased the germinations rates. It has been reported that at non-phytotoxic levels nickel stimulates germination and growth, promotes physiological activities and increased crop yield where as high concentrations prove to inhibitory (Jagetiya and Aery, 1994a). To understand how nickel affects the ability of seed germination and early seedling growth of cowpea, we have examined the comparative response of cowpea (*Vigna unguiculata* L.) under various nickel species.

MATERIALS AND METHODS

Seeds of Vigna unguiculata L. were selected for uniformity, surface sterilized with 0.2% HgCl, solution for five minutes and thoroughly washed for one hour under running tap water. Ten seeds were placed on two layers of Whatman No. 1 filter paper discs in Petri plates which were moistenened with fixed amount of freshly prepared solution of various concentrations of different spices of nickel at regular intervals. A randomized block factorial design with four concentration of each metal species $(1,10,100,1000 \ \mu M)$ was used. For nickel treatment Ni (NO₃)₂ 6 H₂O [E. Merk (India) Limited]; NiCl₂ 6 H₂O [E. Merk (India) Limited, Mumbai] and Ni SO₄ x H₂O [E. Merk (India) Limited, Mumbai] were used. Control sets contained only distilled water. Three replicants for each treatment were kept in dark humid condition at a constant temperature. Experiment was conducted in month of July. Seed germination noted after every two hours. After seven days of treatment, plants were harvested and washed immediately after harvesting. Chlorophyll contents and soluble protein (leaf) contents were measure after Amron (1949) and Bradford (1976), respectively in one-week-old seedlings. Plants were dried in oven at 80°C for 48 hours and weighted for the dry matter production. Following indices were also calculated.

Seedling vigour index (SVI) = Percentage Germination x Hypocotyl length

D	ry weight of control plants-dry weight
	of metal treated plants
Grade of Growth	=X100
Inhibition (GGI)	Dry weight of control plants

RESULTS AND DISCUSSION

Some heavy metals at low doses are essential micronutrient for plants but at higher doses they may cause metabolic disorder and growth inhibition for most of the plant species (Fernandes and Henriques, 1991; Clair *et.al.*, 1991; Thompson *et al.*, 1997; Reeves and Baker, 2000; Raskin and Ensley, 2000). A limited number of publication demonstrated that nickel stimulates germination of some seeds where it is toxic to others. According to Kariev (1969) nickel actually increases the germinating power *Asian J. Bio Sci.* (2007) **2** (1&2)

of the cottonseeds. Jagetiya and Aery (1994 a) and Jagetiya and Bhatt, 2005) observed that at low concentrations NiSO₄ enhances start of seed germination. NiSO₄ solutions at low concentration showed beneficial effect on the germination of lupine (Zanotti, 1938), Soybean (Wu and Yun Hsing, 1958) and barley seeds (Jagetiya and Bhatt, 2005). Certain other studies carried out on the effect of nickel on seed germination and early seedling growth are those of Thukral and Kaur (1987), Ormarod *et al.* (1986), Aggarwal *et al.* (1990), Slivinaskaya (1991), Pandolfini *et al.* (1992), Saradhi and Saradhi (1991), Broderick (1997), Gupta *et al.* (2001), Virginie *et al.* (2005) etc.

Gupta *et al.* (2001) studied the effect of Cu and nickel on seed germination and early seedling growth of *Raphanus sativus* var. Pusa Chetki. He observed that both heavy metals adversely affect seed germination and seedling growth at higher concentration (200 and 500 ppm).

Virginie *et al.* (2005) studied the effect of three-nickel salts (NiCl₂, NiSO₄ and (CH₃Co₂)₂Ni.4H₂O) on germinating seeds of *Gravillea exul* var. rubiginosa an endemic serpentine proteaceae of New Caledonia. They presumed during their investigation that NiCl₂ resulted in the greatest reduction in germination and root growth particularly at 500 mg L⁻¹ followed by NiSO₄ and (CH₂Co₂)₂Ni.4H₂O.

Germination speed and ultimate germination percentage:

In present investigation, in controls as well as in certain doses of nickel, the germination started after 17 hours of installation of the experiment and 100% germination was attained firstly over a 27 hours in 1 μ M Ni (NO₃)₂ and NiCl₂ treatments while in others (NiSO₄), it could be achieved after 29 hours respectively. The higher concentration of different nickel species delayed the germination. After a period of 23 hours germination percentage at 1 μ M nickel concentration was 85% in Ni (NO₃)₂; 70% in NiSO₄ and 80% in NiCl₂, which was higher than control (75%). Ultimate germination percentage was found to be 100%, except in 1000 μ M nickel concentration in all (1,10,100) the treatment of all the three species of nickel. (Table 1).

The seedling vigour index (SVI) was also calculated. The SVI value at 100 μ M nickel treatment were 1924; 1867,1503 for Ni (NO₃)₂; NiSO₄ and NiCl₂, respectively, indicate that NiCl₂ was more toxic and Ni (NO₃)₂ was least toxic for cowpea, the toxicity of different nickel species in respect of seed germination are as follows.

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$$\operatorname{NiCl}_2 > \operatorname{NiSO}_4 > \operatorname{Ni}(\operatorname{NO}_3)_2$$

Nickel chloride wasthe nickel treatment that resulted in the lowest germination. This can be explained by the toxicity of chloride ion, which has been demonstrated in studies of NaCl toxicity in plants (Britto et al., 2004). Nickel sulphate was the least toxic with regard to germination. This may be due to the fact that, unlike the chloride ion, which is a micronutrient, sulphate is a macronutrient and is involved in the synthesis of cell detoxification molecules, such as metallothioneins (Virginie et al., 2005). Curtin et al. (1993) found that sulphate salts induce a batter growth than chloride salt in barley. Higher concentration of nickel declined the activity of acid phosphatase. Lower concentration of nickel stimulates the activity of amylase, protease, acid phosphates and peroxidase in germination of pea, bean, wheat, castor bean, white lupine, soybean, timothy (Mishra and Kar, 1974) and rice (Das et al., 1978). The beneficial effect of nickel on seed germination rates might be related to the activity of urease in germinating seeds and nitrogen economy of developing seedling (Matsumoto et al., 1977 a, b; Sirko and Brodzick, 2000). Urease vis-à-vis nickel appears to have important function in the biochemical process utilizing nitrogen cycle through urea from anabolic reaction during seedling germination and growth (Eskew et al., 1984). Dalton et al., (1988) observed that urease plays an essential role in mobilization of nitrogenous compounds in plants, a process that is especially important during seed germination and fruit formation when protein reserves are degraded in to amino acids. According to urease might function coordinate with arginase in utilizing of seed protein reserve during germination.

Dry Matter Production :

In all the nickel species, lower concentration $(1 \mu M)$ showed an enhancement in shoot-root DMP, which were 12.74 %, 11.11%; 9.80 %, 5.55% and 6.86%, 5.55% for Ni (NO₃) , and NiSO₄ respectively. At 100μ M nickel treatment the reduction in DMP were maximum over controls. Shoot and root showed 11.76%, 30.0%; 14.70%, 47.22% and 16.66%, 48.33% reduction in DMP for Ni (NO₂)₂, NiSO₄ and NiCl₂, respectively. GGI values ranged from 12.74-11. 76, 9.80-14.70, 6.86- 48.33 in shoot and 38.88-30.00, 5.85- 47.22and 5.55-48.33 in root for Ni $(NO_3)_2$, NiSO₄ and Nicl₂, respectively. The effect of nickel on dry matter production is depicted in terms of Grade of Growth inhibition (GGI). On the basis of GGI following toxicity series could be established for cowpea.

> $NiCl_{2} > NiSO_{4} > Ni(NO_{3})_{2}$ HIND AGRI-HORTICULTURAL SOCIETY

rcinol	>		Λ	9	R	2	Rs	
n)	Shoot	Root	Shoot	Root	Shoct	Root	Shoot	Root
itrol	159.05±1.343	45.10 ± 1.55	226.40± 8.20	69.15±1.484	269.50±7.77	81.0 ± 1.41	67.0 ±4.24	15.95 ± 1.060
0	154.50 ± 0.707	42.80 ± 0.282	215.0 ± 7.07	65.50±0.707	257.0 ± 4.24	75.50±0.707	61.0 ±1.41	13.95±0.07
	(-2.86%)	(-5.09%)	(-5.03%)	(-5.27%)	(-4.63%)	(+6.79%)	(-8.95%)	(-12.53%)
0	135.0 ± 7.07	37.10 ± 1.27	197.0 ± 4.24	59.00 ± 4.24	239.50±3.53	69.0 ± 1.41	49.15±1.48	9.50±0.707
	(-15.12%)	(-17.73%)	(-12.98%)	(-1.46%)	(-11.13%)	(-14.81%)	(-26.64%)	(-40.43%)
0	118.0 ± 2.82	30.10 ± 2.68	175.30 ± 2.30	49.50±3.53	217.0±4.24	57.0±4.24	42.05±2.89	8.10=0.424
	(-25.80%)	(-33.25%)	(-22.61%)	(-28.41%)	(-19.48%)	(-29.62%)	(-37.23%)	(-49.2%)

Table 3 : Showing the effect of different nickel species on chlorophyll ('a' 'b' and total) contents (mg g'l) of Vigna unguicallata L.

Nickel		Ni(NO ₃) ₂			NiSO			NiCl ₂	
Concentrations ? M	s Chl. 'a'	Chl. 'b'	Total chl.	Chl. 'a'	Chl. 'b'	Total chl.	Chl. 'a'	Chl. 'b'	Total chl.
Control	0.56 ± 0.014	0.252 ± 0.002	0.826 ± 0.014	0.56 ± 0.014	0.252 ± 0.002	0.826 ± 0.014	0.56 ± 0.014	0.252 ± 0.002	0.826 ± 0.014
	0.584 ± 0.002	0.265 ± 0.004	0.847 ± 0.066	0.573 ± 0.002	0.264 ± 0.002	0.839 ± 0.001	0.571 ± 0.004	0.257 ± 0.004	0.842 ± 0.014
RLH	(4.28%)	(5.15%)	(2.54%)	(2.32%)	(4.76%)	(1.57%)	(1.96%)	(1.98%)	(1.93%)
	0.411 ± 0.001	0.191 ± 0.001	0.554 ± 0.002	0.388 ± 0.011	0.184 ± 0.002	0.53 ± 0.002	0.326 ± 0.001	0.151 ± 0.006	0.48 ± 0.006
сшт	(-26.60%)	(-24.20%)	(-32.92%)	(-30.71%)	(-26.98%)	(-35.83)	(-41.78)	(-40.07)	(41.88%)
00 1 IRA	0.288 ± 0.002	0.136 ± 0.005	0.436 ± 0.004	0.246 ± 0.005	0.133 ± 0.002	0.392 ± 0.005	0.238 ± 0.002	0.131 ± 0.002	0.37 ± 0.002
1.50	(-48.57%)	(-45.03%)	(-47.21)	(-56.07%)	(-47.22%)	(-52.54%)	(-57.5%)	(-48.01%)	(-55.20%)
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Nickel ? M	concentrations	Ni(NO ₃) ₂	NiSO ₄	NiCl ₂
Control		7.2 ± 0.494	7.2 ± 0.494	7.2 ± 0.494
1		7.5 ± 0.212	7.8 ± 0.141	8.00±0.358
		(3.19%)	(3.55%)	(3.92%)
10		8.1 ± 0.282	8.45 ± 0.494	9.1 ± 0.141
		(12.5%)	(17.36%)	(26.38%)
100		10.4 ± 1.31	11.10 ± 0.707	11.7 ± 0.141
		(44.44%)	(54.16%)	(62.5%)
1000		-		-

Table 4 : Effect of different nickel species on soluble protein (leaf) contents (mg g⁻¹) of Vigna unguiculata L.

Saleh (2002) found reduction in dry weight of *Chorcorus olitorius* seedling with increasing nickel concentration with respect to control. Decrease in dry weight at higher concentration of nickel due to the inhibitory effect of heavy metals on cell division and cell elongination and enzyme activity. Growth inhibition caused by nickel can be connected in addition to loss of cellular turgor (Powell *et al.*, 1986; Gabbrielly *et al.*, 1990), also to a reduced extensibility of the cell wall (Fry, 1986; Pandolfini *et al.*, 1992) and it might be due to decreasing efficiency of certain enzymes involved in food and energy utilization (Jagetiya, 1998).

Chlorophyll contents:

Chlorophyll contents are key factor in determining the net plant production. Hence change in chlorophyll content has obvious implication in changes in plant biomass. The results presented in Table 3, indicated that lower level of nickel enhanced the chlorophyll contents while higher concentration proved retardatory. Lower 1µM nickel treatments enhanced the chlorophyll ('a', 'b', and total chl.) contents in all the three species of nickel. Among the three nickel salts the percentage increase over the controls were always higher in Ni (NO₂), in comparison to sulphate and chloride salts. At 100µM nickel addition the decrease in chl.'a' contents over the control was 48.57%; 56.07 % and 57.5 % for Ni $(NO_3)_2$, NiSO₄ and NiCl₂, respectively. The percentage decrease in chl. 'b' was lower than chl.'a' contents. The reduction in chlorophyll 'b' contents over the controls was 46.03; 47.22% and 48.01%, respectively.

According to Pandolfini *et al.* (1992) and Picini and Malvolta (1992), the application of excess levels of nickel caused a marked depression in the chlorophyll contents, whereas intermediate level of nickel it did not vary. These authors suggested that excessive nickel addition probably depress the chlorophyll contents of the leaves by inhibiting the incorporation of Mg in the protoporphyrin molecules. Heavy metals are known to interfere with chlorophyll *Asian J. Bio Sci.* (2007) **2** (1&2) synthesis either through direct inhibition of an enzymatic step or by inducing deficiency of an essential nutrient (van Assche and Clijesters, 1990) and reduction in Hill activity (Basak *et al.*, 2001). Decreased chlorophyll contents associated with heavy metal stress may be the result of inhibition of enzymes responsible for chlorophyll biosynthesis (Zengin and Munzzuroglu, 2005).

Soluble Protein (Leaf) Contents:

Soluble protein (leaf) contents showed increasing trend with increased metal concentration (Table 4). The effect of NiCl₂ on soluble protein (leaf) contents was more pronounced than NiSO₄ and Ni (NO₃)₂. At 100 μ M nickel treatment the percentage increase in soluble protein (leaf) contents were 44.44% for Ni (NO₃)₂, 54.16% for NiSO₄ and 62.50% NiCl₂ (Table 4). Our results favour the findings of Jagetiya and Aery (1994a) and Jagetiya and Bhatt (2005). They all presumed in their investigation that new low molecular weight protein are synthesized that binds with nickel and accelerates breakdown of structural or insoluble proteins. This is a very sensitive and quickchange occurring in the plant system (Dutta, 1979).

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