A Review

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Free radicals, antioxidant activities in fruit crops and their importance as phytomedicines

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Abstract : Free radicals are well documented for playing a dual role in our body both deleterious and beneficial species. In low/moderate concentrations free radicals are involved in normal physiological functions but excess production of free radicals or decrease in antioxidant level leads to oxidative stress. It is a harmful process that can mediates damage to cell structure including lipids, proteins, RNA and DNA, which leads to number of disease conditions. Fruits are good source of natural antioxidants, containing much different antioxidant composition and provide protection against free radicals and causing lower incident and mortality rate of various diseases like cancer and heart diseases in addition to a number of other health benefits. The antioxidant properties of carotenoids are well known and important dietary fruits like orange, tangerine, peach, blueberries and raspberry have good carotenoids content. Citrus fruits and juices contain a unique substance of flavanoids particularly limonene, hesperetin, naringin and narirutin. Cherry contains a high content of total phenols. Pomegranate is considered as super antioxidant because of its antioxidant content and antioxidant activity is higher than other fruits. Interest in the role of fruit antioxidants in human health has promoted research in the field of horticulture. Therefore, it is a demand of modern era to evaluate each fruit in terms of what phytocontituents are present and to use phytocontituents or phytomedicine. In the present article, an attempt has been made to explain the free radicals toxicity, antioxidant properties of fruits and medicinal value of fruits.

Key words : Free radicals, Antioxidants, Oxidative stress, Diseases, Phytoconstituents, Fruits

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The chief toxicity of reactive oxygen species (ROS) resides in their ability to initiate cascade reactions that results in the production of the hydroxyl radicals (.OH) and other destructive species, such as lipid peroxides (LOO.). These cascades are prevented by efficient operations of antioxidant defense systems. The term antioxidant can be considered to describe any compound capable of quenching ROS without itself undergoing conversion to a destructive radical. Organisms have evolved complex enzymatic and non-enzymatic defenses to minimize oxidative damage to macromolecules and cellular structures. They also possess repair system for renewing some oxidative modifications and disposal systems for removing modified macromolecules that are not repaired.

Free radicals are responsible for causing a wide number of health problems such as cancer, aging, heart diseases and gastric problems etc (Bagchi and Puri, 1998; Nagendrappa, 2005). Antioxidant causes protective effect by neutralizing free radicals, which are toxic byproducts of natural cell metabolism. The human body naturally produces antioxidants but the process is not 100 per cent effective in case of overwhelming production of free radicals and that effectiveness also declines with age. Increasing antioxidant intake can prevent diseases and lower the health problems. (Sies, 1991; Goldfarb, 1993). Fruits are loaded with key antioxidants such as vitamin A, C, E, beta-carotene and important minerals, including, selenium and zinc. Fruit research in the last decade has been the most spectacular and intriguing one. Discovery of antioxidant phytochemicals and their promising health promoting effects have paved the way of new revolution what is termed as a fruit revolution. A revolution promising a good health (Chaovanalikit and Wrolstad, 2004). Natural products, mainly obtained from fruits provide a large number of antioxidants. Phytoconstituents are also important source of antioxidant and capable to terminate the free radicals chain (Cody *et al.*, 1986; Oluwaseun and Ganiyu, 2008). In this context, it is of paramount importance to examine varieties, cultivars for their antioxidant content and antioxidant activity for exploring ideal genotypes high in antioxidants.

Fruits are protected against conditions that generate high levels of ROS by a complex antioxidant system which includes three general classes: (1) lipid soluble membrane –associated antioxidants (e.g. β -tocopherol and β carotene); (2) water –soluble reductants (e.g. glutathione and ascorbate); and (3) enzymatic antioxidants (e.g. superoxide dismutase, catalase and peroxidase).Other naturally occurring antioxidants in fruits are also receiving increasing attention. These include isoflavonoids, phenols, polyamines and specific amino acids (Levin *et al.*, 1996; Robards *et al.*, 1999).

A continuous development in understanding relationship between the fruit genotype and diet related diseases have initiated an exciting era for nutritional sciences. Epidemiological studies and neutraceuticals as important tools in promoting health and reducing health care tools. The role of traditional antioxidants such as ascorbic acid and carotenoids has taken a back seat. Phenolic and flavanoids are emerging as ideal candidates for assuming the status vitamins and guardians of health (Mangels et al., 1993). In this context, it is of paramount importance to examine varieties, cultivars for their antioxidant content and antioxidant activity for exploring ideal genotype high in antioxidants. Their exist wide variations that exists among genotypes of a specific crop (Shui and Leong, 2006). This offer a great deal of opportunity for enhancing the levels of antioxidants through genetic manipulation. Higher level of antioxidants in advance breeding lines shows that there is great opportunity to increase the overall antioxidant status of crops by conventional means. Such information could be incorporated into the breeding programmes for breeding antioxidants rich high quality varieties. Natural products, mainly obtained from dietary sources provide a large number of antioxidants. Phytoconstituents are also important source of antioxidant and capable to terminate the free radical chain reactions (Cody et al., 1986; Oluwaseun and Ganiyu, 2008).

Reactive oxygen species (ROS) :

ROS represents a broad category of molecules that indicate the collection of radical and non radical oxygen derivatives (Gul *et al.*, 2000; Dhalla *et al.*, 2000; Irshad and Chaudhuri, 2002; Ray and Husain 2002).

Free radicals:

Molecules or molecular fragment containing single unpaired electron. Thus unpaired electron usually given a considerable degree of chemical reactivity to the radical. In the chemical fragments, the unpaired electron is conventionally shown a superscript "dot" as with the hydroxyl free radical OH (Table 1).

| Table 1 : Few examples of ROS and RNS (Radical and non-radical) | | | | | | |
|-----------------------------------------------------------------|-------------------------------|------------------|----------|--|--|--|
| | Reactive oxygen species (ROS) | | | | | |
| Name | | Chemical | Relative | | | |
| | | formula | activity | | | |
| Radical | Superoxide | O_2 | 0 | | | |
| | Hydroxyl | ·OH | 1 | | | |
| | Hydroperoxyl | HOO [.] | 10^{7} | | | |
| | Alkaxyl | LO [.] | Х | | | |
| | Lipid peroxyl | LOO [.] | | | | |
| Non | Hydrogen peroxide | H_2O_2 | 0 | | | |
| radical | Singlet oxygen | $^{1}O_{2}$ | 0 | | | |
| | Hypochlorous acid | HOCI | 0 | | | |
| | Lipid hydroperoxide | LOOH | Х | | | |

Non- radicals:

Non radicals have no unpaired electron and are generation from free radical species. There intermediates may participate in reactions that give rise to free radicals that damage organic substrate. For example: H_2O_2 (non radical) generated from O⁻ by dismutation of O⁻ anion by SOD (Table 1).

In addition, there is another class of free radicals that are nitrogen derived called reactive nitrogen species (RNS). They are nitrogen derived molecules and are considered a subclass of ROS.

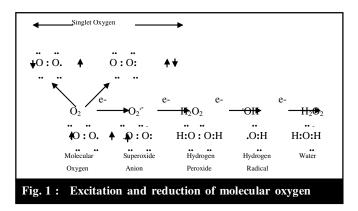
Mechanism of action of ros formation:

In energy producing process of our body, molecular oxygen is reduced by four electrons to yield two molecules of H_2O as the end products. Partially reduced oxygen species or its excited form can be produced (Fig. 1). The molecular (O_2) oxygen is considered to be in a ground (inactive) state and is activated to a singlet (active) state by two different mechanisms;

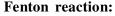
- Absorption of sufficient energy to reverse the spin on one of the unpaired electrons.

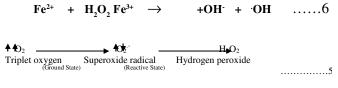
- Monovalent reduction (accepted a single electron): Superoxide radical is formed in the first





monovalent reduction reaction, which undergoes further reduction to form H_2O_2 . H_2O_2 is further reduced to .OH radical in the presence of ferrous State (Fe²⁺). This reaction was first described by Fenton and later developed by Haber and Weiss.





Haber-Weiss reaction:

 $H_2O_2 + OH \rightarrow H_2O + O_2 + H^+ \dots 7$

$$H_2O_2 + O_2^{-} \rightarrow O_2 + OH^{-} + OH \dots 8$$

Sources of ROS production:

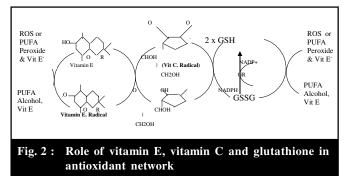
A number of sources of ROS were recognized in physiological and pathological conditions (Fig. 2). Some of the widely accepted sources of ROS are classified in to exogenous and endogenous sources.

- *Exogenous sources include;* Electromagnetic radiation, cosmic radiation, car exhaust, UV light, cigarette smoke, ozone and low wave length electromagnetic radiation.

– Endogenous sources include; Electron leakage during metabolic processes, NADPH oxidase (leukocytes and macrophages), Xanthine / xanthine oxidase aracidonic acid pathway, autooxidation of catecholes and flavonoids and chemicals, atmosphere and environment (bleomycins, paraquat, mineral dust and smog).

Reactive nitrogen species (RNS) :

Nitric oxide (NO) has been the subject of extensive study as a new area of biological research since the



milestone discovery that NO accounts for the biological properties of endothelium-derived relaxing factor (EDRF). Considering that NO has an unpaired electron in its orbit and is thus a free radical and since NO undergoes radical radical reaction with O_2^{-1} it is essential to examine the roles of NO in oxygen radical tissues injury.

| Table 2 : Few examples of RNS (Radical and Non-radical) | | | | | |
|---------------------------------------------------------|---------------------|-------------------------------|----------------------|--|--|
| Reactive nitrogen species (RNS) | | | | | |
| | Name | Chemical formula | Relative activity | | |
| Radical | Nitrosyl cation | NO [.] | Х | | |
| | Nitroxyl anion | NO | Х | | |
| Non | Peroxynitrite | ONOO | Х | | |
| radical | Peroxynitrous acid | ONOOH | Х | | |
| | Nitryl chloride | NO ₂ Cl | Х | | |
| | Nitrogen dioxide | NOO | Х | | |
| | Dinitrogen trioxide | N ₂ O ₃ | Х | | |

Damage by ROS and RNS

Free radicals can disrupt biological systems by damaging a variety of their constituent molecules. Lipids, proteins, carbohydrates and DNA, all are potential targets for the chaotic oxidative attack of radicals produced in their vicinity.

Damage to lipids:

Lipids are most susceptible macromolecules and are present in membrane in the form of polyunsaturated fatty acids (PUFA). ROS attack PUFA in the cell membrane leading to a chain of highly damaging reactions called lipid peroxidation (LPO). These reactions occur in three distinct steps: 1. Initiation 2. Propagation and 3. Termination

Initiation:

The free radical reacts with fatty acid chain (LH) and releases lipid free radical.

$$LH + R \rightarrow L + RH$$

Lipid radical

Propagation:

This lipid radical further reacts with molecular oxygen to form lipidperoxyl radical (LOO). Peroxyl radicals again react with fatty acid to produce lipid free radical.

| $L' + O_2$ | \rightarrow LOO [.] |
|-----------------------|-------------------------------------|
| | Lipid peroxyl radical |
| | |
| LOO [.] + LH | \rightarrow LOOH + L ⁻ |
| Lipid hydroperoxide | |
| | |

 $L^{\cdot} + O_2 \rightarrow LOO^{\cdot}$

Termination:

During termination the two radicals react with each other and the process comes to an end. This process of fatty acid breakdown produces hydrocarbon gases (ethane or pentane).

$$\begin{array}{ccccccc} L & + & L & \longrightarrow & IL \\ LOO & + & LOO & \longrightarrow & LOOL + O_2 \\ LOO & + & L. & \longrightarrow & LOOL \end{array}$$

Under physiological conditions; the most likely fate of carbon centered radicals is to combine with O_2 to generate yet another radical that is peroxyl radical. This Peroxyl radical reacts and attack on adjacent fatty side chain and abstracting hydrogen atom to produce another carbon centered radical and so the chain reaction continues. One .OH radical can result in the conversion of many hundreds of fatty acid chains into lipid hydroperoxides.

 $\begin{array}{cccc} .OH & O_2 & -CH- \\ -CH- & \rightarrow & -C^- & \rightarrow & -COO & \rightarrow & COOH + & -C^- \end{array}$ PUFA Carbon Peroxyl Lipid

UFA Carbon Peroxyi Lipia centered radical radical hydroperoxide

Damage to proteins:

Sulphur containing amino acids having thiol groups specifically are very susceptible to oxidative stress. Activated oxygen can abstract hydrogen atom from cysteine residues to form a thiyl radical that will crosslike to a second thiyl radical to for disulfide bridges. Alternatively, oxygen can add to a methionine residue to form methionine sulphoxide derivatives. Many amino acids undergo specific irreversible modifications when a protein is oxidized. For example, tryptophan is readily cross-linked to form bityrosine products. Oxidative damage can also lead to cleavage of the polypeptide chain and formation of cross-liked protein aggregates. Histidine, lysine, proline, arginine and serine form carbonyl groups on oxidation. The oxidative degradation of protein is enhanced in the presence of metal cofactors that are capable or redox cycling such as Fe.

Damage to DNA:

Activated oxygen and agents that generate oxygen free radicals such as ionizing radiation induce numerous lesions in DNA that lead to deletions, mutations and other lethal genetic effects. Pyrimidine bases are most susceptible to oxidative stress as are purine and deoxyribose sugar. Oxidation of the sugar by the hydroxyl radical is the main cause for DNA strand breaks. Oxidative damage can cause base degradation, DNA fragmentation and cross linking to protein. In addition, incorporation of oxidized deoxyribonucleoside triphosphate causes gene mutation or altered gene expression.

Free radicals in human diseases:

These free radicals have been implicated in the pathogenesis of many human diseases including (Kehrer and Smith, 1994; Gupta *et al.*, 1997; Irshad and Chaudhuri, 2002; O'donovan and Fesnandes, 2004; Agarwal and Prabakaran, 2005; Pourmorad *et al.*, 2006; Dufor *et al.*, 2007; Valko *et al.*, 2007; Pham *et al.*, 2008; Sen *et al.*, 2009):

- Neurodegenerative disorder like Alzheimer's disease, Parkinson's disease, multiple sclerosis, amyotrophic lateral sclerosis, memory loss and depression.

- Cardiovascular disease like atherosclerosis, ischemic heart disease, cardiac hypertrophy, hypertension, shock and trauma.

- Pulmonary disorders like inflammatory lung diseases such as asthma and chronic obstructive pulmonary disease.

- Diseases associated with premature infants, including bronchopulmonary, dysplasia, periventricular leukomalacia, intraventricular hemorrhage, retinopathy of prematurity and necrotizing enterocolitis.

- Autoimmune disease like rheumatoid arthritis.

- Renal disorders like glomerulonephritis and tubulointerstitial nephritis, chronic renal failure, proteinuria, uremia.

- Gastrointestinal diseases like peptic ulcer, inflammatory bowel disease and colitis.

- Tumors and cancer like lung cancer, leukemia, breast, ovary, rectum cancers etc.

- Eye diseases like cataract and age related of ratina, maculopathy.

- Ageing process.
- Diabetes.
- Skin lesions
- Immunodepression.

- Liver disease, pancreatitis.
- AIDS.
- _ Infertility.

Antioxidants defense and its importance in prevention of diseases :

To encounter the harmful affect of ROS/RNS, antioxidant defense mechanism operates to detoxify or scavenge these ROS/RNS. A variety of antioxidant mechanism has been evolved to combat the potential threat of damage to vital biological structures from the aforementioned sources. An antioxidant can be considered as a molecule that, when present at low concentrations compared with those of an oxidizable substrate, significantly inhibits oxidation of that substrate (Simon and Gregory, 1997; Halliwell, 1990; Halliwell and Gutteridge, 1990). This antioxidant affect can be achieved in three different ways as;

Enzyme antioxidants:

Intracellular environment contains enzymes such as superoxide dismutase (SOD), catalase (CAT), glutathione reductase (GR), glutathione peroxidase (GPx), glutathione-S-transferase (GST) and low molecular weight antioxidant glutathione (GSH) that catalyze the breakdown of oxidants generated by cellular metabolism.

Preventative antioxidants:

Preventive antioxidant proteins exist to sequester free transitional metal ions such iron and copper include the iron-binding protein transferring and copper-binding proteins cerruloplasmin and albumin. These antioxidants preventing their interaction with H_2O_2 and O_2^{-} , which would facilitate the production of the produce highly reactive hydroxyl radical (OH).

Sacrificial antioxidant (Chain breaking):

They are powerful electron donors and react preferentially with free radicals before more important target molecules are damaged. In doing so, the antioxidant is sacrificed (oxidized) and must be regenerated or replaced. By, definition, the antioxidant radical is relatively unreactive and unable to attack further molecules.

SOD is an important endogenous antioxidant enzyme act as the first line of defense system against ROS which scavenges superoxide radicals to H₂O₂. GPx present in the cytoplasm of the cells removes H_2O_2 by coupling its reduction to H₂O with oxidation of GSH. GR is a flavoprotein enzyme, regenerates GSH from oxidized glutathione (GSSG) in the presence of NADPH. GSH is a tripeptide and a powerful antioxidant present within the cytosol of the cells and is the major intracellular nonprotein thiol compound. SH groups present in GSH to react with H₂O₂ and .OH radical and prevent tissue damage and GSH is also capable of scavenging ROS directly or enzymatically via GPx. Vitamin C and E are non enzymatic endogenous antioxidants also exists within normal cells and react with free radicals to form radicals themselves which are less reactive than the radicals. They break radical chain reactions by trapping peroxyl and other reactive radicals (Valko et al., 2006). Non enzymatic antioxidants also can be divided into metabolic antioxidants and nutrient antioxidants. Metabolic antioxidants are the endogenous antioxidants which are produced by metabolism in the body like lipoid acid, glutathione, L-arginine, coenzyme Q10, melatonin, uric acid, bilirubine, metal-chelating proteins, transferring etc. (Nagendrappa, 2005; Pham-Huy et al., 2008). While nutrient antioxidants belonging to exogenous antioxidants, which cannot be produced in the body but provided through diet or supplements viz., trace metals (selenium, manganese, zinc), flavonoids, omega-3 and omega-6 fatty acids etc. (Goldfarb, 1993; Carr and Frei, 1999)). In spite of these mechanisms free radicals are constantly generated. For this reason, body fluids contain a rich array of low molecular weight molecules that preferentially reacts with free radicals before more important structures are damaged but are sacrificed in the process. These can be conventionally divided into those that are water soluble and those that are lipid soluble and exit in environment such as lipoproteins and cell membranes to prevent the propagation phase of lipid peroxidation (chain breaking antioxidants). Of the aqueous molecules, the best known is vitamin C (ascorbate), which is the most powerful electron donor and is the first plasma antioxidant to be sacrificed upon exposure to oxidative stress. Its oxidation product dehydroascorbate can then be regenerated to ascorbate intracellularly. Man is unable to syntheses vitamin C for which fresh fruits and vegetables form the major dietary sources. Whilst the more recently recognized group of antioxidants are the polyphenols including flavonoids which are active free radical scavengers. Their dietary sources are fruits and vegetables. Vitamin E is a lipid soluble antioxidant and refers to alpha-tocopherol and other related tocotrienols. Important dietary sources are vegetable oils, cereals grains etc. It is the major antioxidant of lipoproteins and membranes where it acts a powerful chain breaking antioxidant. The efficient function of vitamin E depends upon its regeneration following oxidation to the tocopheryl radical. This is achieved by a direct electron transfer with vitamin C at the lipid water interface of lipoproteins. Beta

carotene is the most abundant carotenoid and circulates in lipoproteins. Its major dietary sources are fruits and vegetables. Beta carotene quenches singlet oxygen but is nota particularly active radical scavenger other than at low oxygen tensions.

Antioxidants in fruits and their medicinal values :

Epidemiological studies have shown that consumption of fruits have health benefits against chronic diseases including cardiovascular disease and certain types of cancer (Doll, 1990; Hertog et al., 1997; Knekt et al., 1996; Rimm et al., 1996). The health-promoting properties of fruits are due to the presence of some vitamins (A, C, E, and folates), dietary fiber, and nonessential phytochemicals in these food products. Among phytochemicals, polyphenols deserve a special mention due to their free radical scavenging activities and in vivo biological activities that are being investigated by many researchers (Rimm et al., 1996, Bors et al., 1990, Chen et al., 1996, Rice-Evans et al., 1996, Bravo et al., 1998).

Citrus:

Epidemiological evidence has suggested that consumption of fruit reduces the risk of both cancer and cardiovascular diseases, potentially through the biological actions of components such as vitamin C, vitamin E, flavonoids and carotenoids (Gey et al., 1991; Ziegler, 1989; Gaziano et al., 1992). Citrus species are extremely rich sources in vitamin C and flavanones, a class of compounds which belongs to the flavonoids family. Lemon, lime, mandarin and sweet orange contain essentially rutinosides, mainly of hesperetin, while grapefruit and bitter orange contain neohesperidosides, mainly of naringenin. Some varieties of sweet orange are also characterised by a high content of anthocyanins, a class of flavonoids widely used as natural colorants for food products. These compounds are present both in the juices and the tissues of fruit; thus, citrus fruit and their derived products represent a major source of flavanones and other flavonoids and may contribute significantly to the total dietary flavonoid intake in humans (Tomas and Clifford, 2000). Cholesterol-lowering effects by citrus flavonoids have also been demonstrated in cultured liver cells (HepG2) and they were associated with a reduced synthesis of hepatic cholesterol esters through regulation of apolipoprotein B metabolism (Borradaile et al., 1999). Others have shown a 22 per cent tumor incidence reduction in male rats fed orange juice after induction of colon cancer by azoxymethane (Miyagi et al., 2000). Furthermore, we have recently shown that hesperetin glucuronides, potential in vivo metabolites of the citrus flavonoid hesperidin, are able to protect human skin fibroblasts from UVA-induced oxidative stress (Proteggente et al., 2003).

Olive:

Nowadays, cardiovascular diseases are still the main health concern in developed countries. However, the incidence is very different among developed countries, being the Mediterranean area, the one with the lowest rate (Sans et al., 1997; Tunstall-Podoe et al., 1999; Covas et al., 2000). Mediterranean countries consume high amounts of fat (at least 35 per cent of consumed energy comes from fat) mainly as monounsaturated fatty acids (MUFA) present in olive oil (Serramajem, 1995). Consumption of virgin olive oil, a fat that is characterized for its rich composition in MUFA and antioxidants (vitamin E, â- carotene and phenolic compounds) could decrease the susceptibility of LDL to oxidation (Mata et al., 1996; Nicolaiew et al., 1998; Ramírez et al., 1999; Gimeno et al., 2002), possibly due to its enrichment in oleic acid and in minor compounds, present in the diet, that may bind to LDL. Until now the effect of virgin olive oil phenolic compounds in LDL has been studied in vitro and in vivo evaluating the improvement of LDL resistance to oxidation (Caruso et al., 1999, Visioli and Galli, 1994, Visioli et al., 1995).

Grapes:

Grapes, Vitis vinifera L. (Family: Vitaceae) are one of the most widely consumed fruits all over the world and have considerable importance for their medicinal and nutritive value for thousands of years. Traditionally, grapes were used to treat throat infection and dried grapes (raisins) to treat constipation, skin, kidney, and liver diseases. Grapes, wine and other grape products are rich sources of antioxidants with a high potential for the prevention and treatment of human malignancies. The key components responsible for these beneficial effects are widely considered to be complex phenols, which have been reported to possess various potent biological activities such as antioxidant, antiviral, enzyme inhibiting, antitumor and anti-HIV activities (Iacopini et al., 2008, Iijima et al., 2002; Lange, 2007). The polyphenols of grape seeds exhibit a strong antioxidant activity. Scientific studies have claimed that the antioxidant potential of grape seeds proanthocyanidins is 20 times greater than vitamin E and 50 times greater than vitamin C (Bagchi et al., 1997). Recent studies have shown that procyanidins in grape seeds possess anti-inflammatory, anti-arthritic, anti-allergic effects, prevent heart diseases and skin aging (Maffei et

al., 1996) and improvement in vision performance (Corbe et al., 1988). Crushed grape seeds are the source of grape seed oil. Grape seed oil contains nutritionally useful essential fatty acids and tocopherols (vitamin E) (El-Mallah et al., 1993). GSE (Proanthocyanidin) extract have been marketed in France for decades as treatment for venous and capillary disorders (e.g., retinopathies, venous insufficiency, and vascular fragility) (Murray and Pizzorno, 1998).

Almond:

The proximate composition of almond includes 50.6 per cent lipid, 21.3 per cent protein, 19.7 per cent carbohydrate, 5.3 per cent water, and 3.1 per cent ash (w/w) (USDA, 2004). Almond oil is also a rich source of a-tocopherol (around 390 mg/kg) and contains trace amounts of other tocopherol isomers as well as phylloquinone (70 mg/kg) (USDA, 2004). Almond oil contains 2.6 g/kg phytosterols, mainly bsitosterol with trace amounts of stigmasterol and campesterol (USDA, 2004). The compositional characteristics of almond oil show that it is rich in several health-promoting nutrients, many of which may be responsible for the observed beneficial effects of dietary almond consumption in cardiovascular diseases (Sabate and Fraser, 1994) and in weight management (Fraser et al., 2002). Several studies have reported that almond consumption may improve blood lipid profiles by lowering low-density lipoprotein (LDL) cholesterol and raising plasma high-density lipoprotein (HDL) cholesterol levels. Thus, there is much current interest in almond oil as health-promoting edible oil (Spiller et al., 1998). The extracted oil, which is eatable and has a nutty taste, is used to clean the skin from spots.

Berries:

Few fruits have quite the provocative allure, the fragile charm or the nutrients of berries. They are full of fiber, minerals and vitamins, and loaded with healing antioxidants. Blueberries, raspberries and blackberries are rich in proanthocyanidins, antioxidants that can help prevent cancer and heart disease. Strawberries, raspberries and blackberries contain ellagic acid, a plant compound that combats carcinogens. Blueberries also appear to delay the onset of age-related loss of cognitive function. Strawberries have been shown to be a rich source of phenolic compounds with antioxidant and antiproliferative activities (Halvorsen et al., 2002; Wang et al., 1996; Guo et al., 2003; Sun et al., 2002; Meyers et al., 2003). It has further been shown that total antioxidant activity in serum from elderly women increased after consumption of strawberries (Cao et al., 1998). The antioxidant properties of strawberries have been demonstrated to be mainly due to high content of phenolic compounds rather than vitamin C (Guo et al., 2003; Sun et al., 2002). Strawberries, together with raspberries and blackberries, are the main sources of ellagic acid in the human diet (Daniel et al., 1989; Clifford and Scalbert, 2000). The highest level of total ellagic acid was found in strawberry leaves followed by achenes and finally flesh (Maas et al., 1991). Edible blueberry species are well recognized for their potential health benefits. By use of HPLC-PDA, the phenolics constituents in the berries were identified as chlorogenic acid, p-coumaric acid, hyperoside, quercetin-3-O-glucoside, isoorientin, isovitexin, orientin and vitexin (Dastmalchi et al., 2011).

Apples:

Apples contain a large concentration of flavonoids, as well as a variety of other phytochemicals. The compounds most commonly found in apple peels consist of the procyanidins, catechin, epicatechin, chlorogenic acid, phloridzin, and the quercetin conjugates. In the apple flesh, there is some catechin, procyanidin, epicatechin, and phloridzin, but these compounds are found in much lower concentrations than in the peels (Escarpa and Gonzalez, 1998). A reduced risk of cardiovascular disease has been associated with apple consumption. The Women's Health Study surveyed nearly 40,000 women, and women ingesting the highest amounts of flavonoids had a 35 per cent reduction in risk of cardiovascular events (Sesso et al., 2003). Women ingesting the highest amounts of flavonoids had a 35 per cent reduction in risk of cardiovascular events. Apple consumption has been inversely linked with asthma and has also been positively associated with general pulmonary health. In a recent study involving 1600 adults in Australia, apple and pear intake was associated with a decreased risk of asthma and a decrease in bronchial hypersensitivity, but total fruit and vegetable intake was not associated with asthma risk or severity (Woods et al., 2003). Apple consumption may also be associated with a lower risk for diabetes. In Finnish study of 10,000 people, a reduced risk of Type II diabetes was associated with apple consumption (Knekt et al., 2002). Apple intake has also been associated with weight loss in middle aged overweight women in Brazil (de Oliviera et al., 2003).

Peach:

The phytochemicals reported in Prunus L. include carotenoids, anthocyanins, and other phenolics (Cevallos-Casals et al., 2005; Gao and Mazza, 1995; Gil et al., 2002; Radi et al., 1997; Senter and Callahan, 1999; Tourjee et

al., 1998; Weinert *et al.*, 1990; Werner *et al.*, 1998). Orange-fleshed peaches have the carotenoids, carotene and cryptoxanthin (Tourjee *et al.*, 1998). Several hydroxycinnamates such as chlorogenic acid and neochlorogenic acid, flavan 3-ols such as catechin and epicatechin, and flavonols such as quercetin 3-rutinoside have been identified in peaches and plums (Kim *et al.*, 2003; Tomas-Basbenin, 2001). Anthocyanins and other phenolic compounds are responsible for many health benefits related to cancer prevention and cardiovascular health (Edenharder *et al.*, 2003; Moline *et al.*, 2000; Sun *et al.*, 2002; Wang *et al.*, 1997, 1999; Zhou *et al.*, 2004).

Walnut:

Walnuts contain about 65 per cent lipids, however, considerable differences exist among varieties (range: 52-70%, w/w) (USDA, 2004; Zwarts et al., 1999). Walnuts also contain 15.8 per cent protein, 13.7 per cent carbohydrate, 4.1 per cent water, and 1.8 per cent ash (w/w) (USDA, 2004). Evidence from epidemiologic and intervention studies as well as clinical trials shows that walnut consumption has favourable effects on serum lipid levels in humans, such as lowering LDL, raising HDL, and reducing total serum triacylglycerol levels, all of which reduce the likelihood of suffering from a cardiovascular event (Abbey et al., 1994; Sabate et al., 1993; Zibaeenezhad et al., 2003). The antioxidative components of walnut oil have significant antiradical properties that may exert a protective effect against the oxidation of biomacromolecules such as LDL, a known risk factor for atheroma development and, thus, heart disease. (Espin et al., 2000)

Pomegranate:

It is one of the important commercial horticultural fruits which is generally very well adapted to the Mediterranean climate (Biale, 1981). Pomegranate fruits are consumed fresh or processed as juice, jellies and syrup for industrial production (Hodgson, 1917; La Rue, 1969; Nagy *et al.*, 1990). The edible part of the fruit contains considerable amounts of acids, sugars, vitamins, polysaccharides, polyphenols (ellagic acid, punicalagin etc.) and important minerals (Gil *et al.*, 2000; Kulkarni *et al.*, 2004). Different parts of its tree (leaves, fruits and bark skin) have been used traditionally for their medicinal properties and for other purposes such as in tanning (Rania *et al.*, 2008). It is proved to have high antioxidant activity (Gil *et al.*, 2003).

Jamun:

Syzygium cumini, widely known as Jamun, is a tropical tree that yields purple ovoid fleshy fruit. Its seed has traditionally been used in India for the treatment of diabetes (Benherlal and Arumughan, 2007). Jamun is the best nutraceutical fruit with natural curing and food industries for manufacturing commercially viable food products (Muhammad et al., 2009). Jamun fruit is one of those which contain a variety of important nutritional compositions. For instance, the fruit syrup is very useful for curing diarrhea. It is stomachache, carminative and diuretic, apart from having cooling and digestive properties (Thaper, 1958). Vitamin C is one of the most crucial vitamins in human that plays a large role in hundreds of the body's functions and we must obtain vitamin C through our diet particularly from fruits like jamun. Vitamin C may also be able to regenerate other antioxidants such as vitamin E (Carr and Frei, 1999 and Simon and Hudes, 2000).

Guava:

Psidium guajava or guava is very rich in antioxidants and vitamins. For every 100 gram of guava, there are 59.5 ig carotene, 9.9 ig retinol equivalent, 0.1 mg vitamin B1, 0.05 mg vitamin B2, 1.08 mg niacin, 6.76 g fibre and 151.4 mg vitamin C (ascorbic acid). Guava is also high in lutein, zeaxanthine and lycopene (Tee et al., 1997; Hobert and Tietze, 1998) research reported that guava consumption in hypertensive and hypercholesterolemia are able to increase significantly the intake of carbohydrate, fibre, potassium and vitamin C. The increase of vitamin C intake was shown to be associated with increase of plasma ascorbic acid (45.3 %), high density lipoprotein-cholesterol (HDLcholesterol) (8.9 %) and decrease of total cholesterol (11.9 %), triglyceride (8.1 %), systolic (5.0 %) and diastolic (4.6 %) (Singh and Rastogi, 1997).

Phytomedicine as antioxidant:

Recent researches have shown that the antioxidants of plant origin with free-radical scavenging properties could have great importance as therapeutic agents in several diseases caused due to oxidative stress ((Pham-Huy *et al.*, 2008; Valko *et al.*, 2007; Agarwal and Prabakaran, 2005; Pourmorad *et al.*, 2006; O'donovan and Fernandes, 2004; Dufor *et al.*, 2007; Sen *et al.*, 2009). Plant extracts and phytoconstituents found effective as radical scavengers and inhibitors of lipid peroxidation (Dash *et al.*, 2007; Yildirim *et al.*, 2001). Many synthetic antioxidant compounds have shown toxic and/or mutagenic effects, which have stimulated the interest of many investigators to search natural antioxidant (Nagulendran *et al.*, 2007). The use of phytoconstituents as drug therapy to scavenge free radicals and to treat disorders leads due to oxidative stress has proved to be clinically effective and relatively less toxic than the existing drugs. Therefore, it is demand of time to uses drugs from fruit sources or phytoconstituents to prevent and/or treat oxidative stress. The list of different phytochemicals present in fruits given in Table 3 and some fruits producing antioxidant activity are enlisted in Table 4.

Conclusion and directions for future research:

Currently, there has been an increased interest globally to identify antioxidant compounds in fruits which are pharmacologically potent and have low or no side effects for use in protective medicine and food industry. In the modern world, the use of different chemicals, pesticides, pollutant, smoking and alcohol intake and even some of synthetic medicines increases the incidences of chronic disease due to overproduction of free radicals. Fruits produces large amount of antioxidants to prevent the oxidative stress, they represent a potential source of new compounds with antioxidant activity. More or less the free radicals plays a role in health of modern era and the diseases caused from free radical are becoming a part of normal life. Increasing knowledge in antioxidant phytoconstituents and include them in daily uses and diet can give sufficient support to human body to fight against these diseases. Explore the antioxidant principles from fruit resources, identification and isolation of phytoconstituents and simultaneously presenting enormous scope for their better therapeutic application for treatment of human disease. Therefore, it is time for us, to explore and identify our traditional therapeutic knowledge and fruit sources and interprets them according to the recent advancements to fight against oxidative stress, in order to give it a deserving place. Molecular techniques can make a major contribution to this area of research. Specific probes for the genes of the different isoforms can help determine which are affected by the conditions imposed on the fruits. However, although transcript for specific isoforms can be quantified,

| Table 3 : Major phytochemicals in fruits | | | |
|------------------------------------------|-----------------------------------------------------------------------------------|--|--|
| Phytochemicals | Fruits | | |
| Terpenes | Carotenoids and Limnoids | | |
| Polyphenolics | Anthocyanidins, Catechins, Hesperetin, Naringin, Rutin, Quercetin and Punicalagin | | |
| Organosulfur compounds | Glucosinolates, Indoles and Allylic sulfides | | |
| Phenolic acids | Ellagic acid and Chlorogenic acid | | |

| Table 4 : Phytoconstituents with antioxidant activity | | | | |
|-------------------------------------------------------|----------------------------------------------------------------------------------|-----------------------------------------|--|--|
| Phytoconstituents | Examples | Common fruit sources | | |
| Anthocyanidins | Cyanidin, Delphinidin, Malvidin, Pelargonidin, Peonidin, | Red, blue and purple berries; red and | | |
| | Petunidin | purple grapes; red wine | | |
| Flavanols | Monomers (Catechins) : Catechin, Epicatechin, | Catechins (grapes, berries, apples) | | |
| | Epigallocatechin, Epicatechin gallate, Epigallocatechin gallate | | | |
| | Dimers and polymers (Theaflavins, thearubigins, | Proanthocyanidins (apples, berries, red | | |
| | Proanthocyanidins) | grapes, red wine) | | |
| Flavanones | Hesperetin, Naringenin, Limonene, Tangeretin, Nobelitin | Citrus fruits and juices, e.g. oranges, | | |
| | | grapefruits, lemons | | |
| Flavonols | Quercetin, Kaempferol, Myricetin, Isorhamnetin | Widely distributed (apples, berries) | | |
| Flavones | Apigenin, Luteolin | Berreis | | |
| Carotenoids | β-carotene,Lutein,Lycopene | Papaya, mango, peach, prunes and | | |
| | | orange | | |
| Tannis | Ellagitannins and propelargonidin | Juman (Syzygium cumini) | | |
| Stilbenes | Cajaninstilbene acid | Cajanus cajan | | |
| Phenolic acids | Capssaicin, Ellagic acid, Gallic acid, Rosmarinic acid and | Pomegranates, walnuts, red grapes (and | | |
| | Tannic acid. | red wine and raisins), cocoa, berries | | |
| Volatile and essential oils | α -terpinene, α -pinene, myrcene, p-cymene, β -phellandrene, | Citrus reticulate | | |
| | citronellol, trans-geraniol, a-copaene, agarospirol | | | |
| Hydrocinnamic acids | Caffeic acid, Chlorogenic acid, Frulic acid and p- Coumaric acid | Apple,pear,peach,apricot,mango,avocad | | |
| | | o,grapes. | | |

increases in transcript abundance may not necessarily be accompanied by increase in enzyme activities. Therefore, the research should be a combination of determination of both gene expression and enzyme activity. Only after the specific genes have been targeted can breeding for improved storage be investigated.

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